

HIGHLY OXYGENATED GUAIANOLIDES FROM *OTANTHUS MARITIMUS*

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Key Word Index—*Otanthus maritimus*; *Pentzia albid*a, Compositae, sesquiterpene lactones, guaianolides; seco-guaianolide, monoterpenes

Abstract—The reinvestigation of the aerial parts of *Otanthus maritimus* afforded in addition to five known guaianolides 33 new ones, a seco-guaianolide and three monoterpene diols. The structures were elucidated by high field NMR spectroscopy and chemical transformations. The aerial parts of *Pentzia albid*a gave a further guaianolide closely related to one of the types from *Otanthus*.

INTRODUCTION

The monotypic genus *Otanthus* is widespread in the coastal areas of the Mediterranean. *O. maritimus* (L.) Hoffm. et Link. (Arabic name, Gaadeh) is used by the bedouins for treating asthmatic bronchitis. It has been studied chemically by different groups. The roots contained acetylenes and highly unsaturated amides [1] while from the aerial parts several widespread terpenoids were reported [2, 3]. Furthermore a guaianolide [4] and flavone glycoside [5] were isolated. A reinvestigation of the aerial parts afforded a highly complex mixture of oxygenated guaianolides. The results are discussed in this paper.

RESULTS AND DISCUSSION

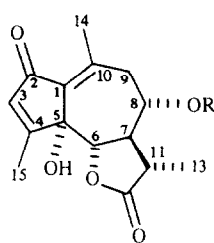
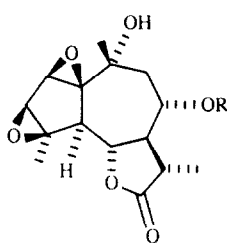
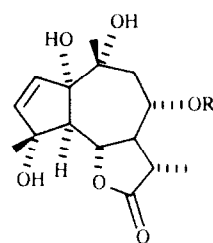
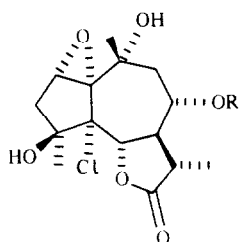
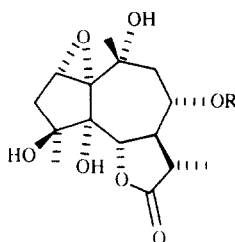
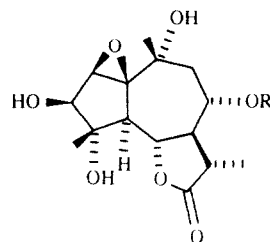
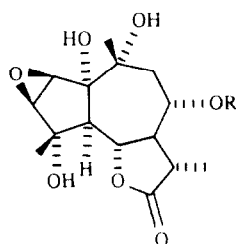
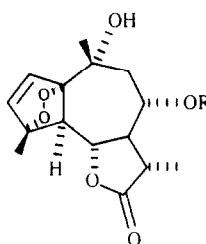
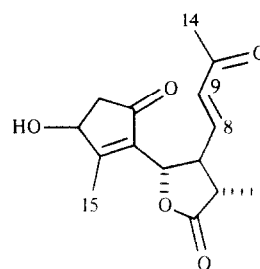
The polar fraction of an extract of the aerial parts of *O. maritimus* afforded a complex mixture of sesquiterpene lactones. Finally in addition to artabin [6] the guaianolides **1a–d**, **2a–d**, **3a–d**, **4a–d**, **5a–d**, **6a–d**, **7a–e** and **8a–d**, the seco-guaianolide **10**, the monoterpene diols **11a/b** and **12** as well as the known ionone derivatives romalea allene [7] and vomifolol [8], undeca-2*E*,4*E*-diene-8,10-diynoic acid isobutylamide [1], liliolide and sesamin were found. A sample of the same species grown in the Botanical Garden Berlin gave tanaparthin-1*α*,4*α*-peroxide [9], **9a** and **9b–d** [10].

The ¹H NMR spectrum of **1a** (Table 1) showed that an angelate was present. Furthermore signals for two olefinic methyls at δ 2.43 and 2.28 as well as for an olefinic proton (δ 6.05) indicated the presence of a derivative of desacetyl matricarin. As the H-6 signal was a doublet, a 5-hydroxy group was very likely. The down field shift of H-7 (δ 3.11) and H-9*α* (δ 3.25) required a 5*α*-orientation of the hydroxy group. The remaining signals could be assigned by spin decoupling. The couplings of H-8 indicated the *α*-position of the angelate residue. The ¹H NMR data of **1b–d** (Table 1) indicated the presence of the corresponding isobutyrate, isovalerate and methylbutyrate. The latter two esters could not be separated as in all the other cases where the same mixture of esters were present (**2c/d–9c/d**).

The ¹H NMR spectra of **2a–d** (Table 1) were close to that of artecannin [11] and those of **3a–d** to that of canin. However, the presence of 11*β*,13-dihydro derivatives followed from the replacement of the exomethylene signal by methyl doublets and a double quartet for H-11, its coupling indicating 11*α*-methyl compounds. Furthermore the presence of 8*α*-acyloxy derivatives could be deduced from the threefold doublets around δ 5.2. The nature of the ester groups followed from the typical signals. One of the lactones, the methylbutyrate **3b** was isolated previously from the same species [4], but the stereochemistry has to be revised.

The ¹H NMR spectra of **4a–d** (Table 1) again showed that these lactones only differed in the nature of the ester groups. In the spectrum of **4b** in deuteriomethanol all signals could be assigned by spin decoupling. The low field pair of doublets with a 5.5 Hz coupling clearly indicated the presence of a guaianolide with a 2,3-double bond. As no additional couplings were present no hydrogens were at C-1 and C-4. As followed from the molecular formula (C₁₉H₂₈O₇) most likely three hydroxy groups were present. The similarity of the remaining signals with those of **3b** indicated an identical structure of the seven-membered ring. The stereochemistry was determined by NOE difference spectroscopy. Thus clear effects were observed between H-14, H-15 (5%), H-2 (5%), H-6 (10%), H-8 (6%) and H-9*β* (4%), between H-15, H-14 (5%), H-6 (12%) and H-3 (5%), between H-13 and H-7 (6%) as well as between H-8, H-11 (5%) and H-6 (6%). The ¹³C NMR spectrum (Table 2) also supported the structure. The ¹H NMR spectrum of **4a** showed that the corresponding angelate was present while the unseparable mixture of **4c/d** was the isovalerate and the 2-methylbutyrate (Table 1). A 8-desacyloxy derivative has been isolated previously [12]. The ¹H NMR data are very close to those of **4a–d**.

In the mass spectrum of **5b** the highest very weak ion corresponds to C₁₉H₂₇O₇Cl and high resolution of *m/z* 366 showed that this fragment was due to C₁₉H₂₆O₇. Accordingly, **5b** most likely was a chlorohydrin. Again the ¹H NMR spectrum (Table 1) was in part close to that

**1 a-d****2 a-d****3 a-d** (bis α -epoxide)**4 a-d****5 a-d****6 a-d****7 a-f****8 a-d****9 a-d****10****1a-9a** R = Ang, **1b-9b** R = *t*Bu, **1c-9c** R = *t*Val, **1d-9d** R = MeBu, **7e** R = Prop**7f** 8-desacloxy - 11,13-dehydro, **7g** R = Ang, 3-OAc, **7h** R = Ang, 10-OAc, **7i** R = Ang, 3,10-OAc

of **3b**. However, the H-3 signal was replaced by two signals at δ 2.14 (*d*) and 1.69 (*dd*) and the H-2 signal was slightly shifted down field (δ 3.48). The latter showed only a small coupling with H-3' (3 Hz). This is typical for epoxy protons. All data therefore agreed with the structure **5b**. The stereochemistry could not be deduced from the data. This, however, was possible by NOE difference spectroscopy. Clear effects were observed between H-14, H-6 (10%), H-8 (6%) and H-2 (6%), between H-6, H-14 (5%) and H-8 (3%) as well as between H-13 and H-7 (10%) while no effect between H-15 and H-6 was obtained.

The ^1H NMR spectra of **5a** and **5c/d** (Table 1) showed that again the corresponding angelate, isovalerate and methylbutyrate were present.

The ^1H NMR spectrum of **6b** (Table 1) was very close

to that of **5b**. Small changes, especially a down field shift of the H-7 signal indicated that the corresponding 5 α -hydroxy derivative was present. This was supported by W-couplings of H-14 with H-9 α and 10-OH, most likely due to a fixed conformation by a hydrogen bond between 5 α - and 10 α -hydroxy (OH δ 2.49 *br.s*). As the chemical shifts and the couplings were nearly identical in the case of **5b** and **6b** the stereochemistry surely also was the same. The spectrum of **6a** and **6c/d** (Table 1) indicated that the corresponding angelate, isovalerate and methylbutyrate also were present.

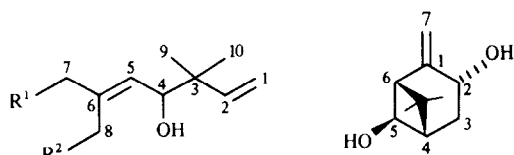
The ^1H NMR spectra of **7a-e** (Table 3) again showed that these lactones only differed in the nature of the ester groups. In the spectrum of the angelate **7a** all signals could be assigned by spin decoupling. As followed from the molecular formula ($\text{C}_{20}\text{H}_{28}\text{O}_8$) an epoxytriol angel-

Table 1 ^1H NMR spectral data of **1a/b**, **2a/b**, **3a/b**, **4a/b**, **5a/b** and **6a/b** (400 MHz, CDCl_3 , δ -values*)

H	1a	1b	2a	2b†	3a	3b†	4a	4b†‡	5a	5b†	6a	6b†
2	—	—	3.60 br s	3.61 br s	3.46 d	3.45 d	5.96 d	5.84 d	3.86 d	3.84 d	3.77 d	3.75 d
3	6.05 dq	6.05 dq	3.28 br s	3.28 br s	3.29 d	3.28 d	5.91 d	5.81 d	{ 2.15 d 1.73 dd	{ 2.14 d 1.69 dd	{ 2.19 d 1.76 dd	{ 2.19 d 1.76 dd
5	—	—	2.80 d	2.78 d	2.58 d	2.56 d	2.61 d	2.46 d	—	—	—	—
6	3.98 dd	3.96 dd	4.27 t	4.23 t	4.18 dd	4.15 dd	4.39 dd	4.61 dd	4.92 d	4.87 d	4.85 d	4.81 d
7	3.11 q	3.10 q	2.83 q	2.82 q	3.08 q	3.03 q	2.58 q	2.51 q	2.33 q	2.31 q	2.57 m	2.58 m
8	4.97 ddd	4.81 ddd	5.29 ddd	5.17 ddd	5.25 ddd	5.11 ddd	5.16 ddd	5.04 ddd	5.37 dt	5.19 dt	5.29 ddd	5.13 ddd
9 α	3.25 dd	3.18 dd	2.38 m	2.36 m	2.18 dd	2.15 dd	2.42 t	2.43 t	2.03 dd	1.95 dd	2.00 t	1.97 t
9 β	2.25 dd	2.16 dd	2.15 m	2.14 m	2.10 dd	2.07 dd	1.92 dd	1.78 dd	2.32 dd	2.26 dd	2.16 dd	2.15 dd
11	2.53 dq	2.52 dq	2.55 dq	2.49 dq	2.52 dq	2.48 dq	2.58 m	2.63 dq	2.71 dq	2.64 dq	2.56 m	2.56 m
13	1.38 d	1.38 d	1.28 d	1.33 d	1.30 d	1.32 d	1.36 d	1.30 d	1.33 d	1.35 d	1.34 d	1.33 d
14	2.43 s	2.42 s	1.12 s	1.13 s	1.16 s	1.12 s	1.19 s	1.15 s	1.56 s	1.53 s	1.57 s	1.54 s
15	2.28 d	2.28 d	1.53 s	1.52 s	1.54 s	1.52 s	1.44 s	1.37 s	1.58 s		1.60 s	1.59 s

*OAng 6.15–6.21 qq, 2.04–1.99 dq, 1.92–1.89 dq, OMeBu⁺ 2.43–2.49 tq, 0.94–0.91 t, 1.21–1.18 d, OrVal 2.18–2.22 d, 1.00–0.97 d, OrBu. 2.58–2.55 qq, 1.20 d, OProp 2.40 ABM₃, 1.20 t, †Isovalerates and methylbutyrates nearly identical signals, ‡D₃COD

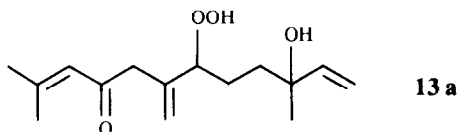
$J[\text{Hz}]$ Compounds **1a–d** 3,15=1.5, 6,7=7.8=7.11=8.9=11, 8,9'=2, 11,13=7, 9,9'=15, compounds **2a–d** 5,6=6,7=7.8=10, 7,11=12, 8,9=8,9'=7, 11,13=7, compounds **3a–d** 2,3=1, 5,6=11, 6,7=10, 7,8=10, 7,11=11, 8,9=4, 8,9'=5, 9,9'=15, compounds **4a–d** 2,3=5.5, 5,6=11, 6,7=7.8=7.11~10, 8,9=9,9'=11, 8,9'=4, 11,13=7, compounds **5a–d** and **6a–d** 2,3'=3, 3,3'=16.5, 6,7=10, 7,8=7.11=8,9'=11, 8,9=3.5, 9,9'=13, 11,13=7



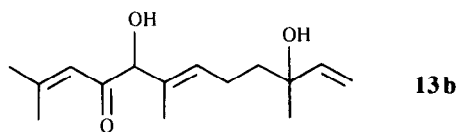
11a $\text{R}^1 = \text{H}$ $\text{R}^2 = \text{OH}$

11b $\text{R}^1 = \text{OH}$ $\text{R}^2 = \text{H}$

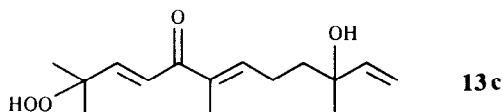
12



13a



13b



13c

Table 2 ^{13}C NMR spectral data of **4b**, **7a** and **7f** (100.6 MHz, CDCl_3 , δ -values*)

C	4b†	7a‡	7f
1	90.2 s	72.9 s	73.3 s
2	135.2 d	64.2 d	64.4 d
3	140.7 d	64.4 d	64.7 d
4	83.5 s	83.1 s	72.2 s
5	52.8 d	57.3 d	57.9 d
6	75.0 d	76.5 d	79.9 d
7	43.9 d	41.5 d	45.1 d
8	66.1 d	72.2 d	23.6 t
9	41.2 t	43.2 t	35.1 t
10	79.3 s	71.5 s	83.4 s
11	50.6 d	51.6 d	139.3 s
12	177.7 s	177.2 s	169.4 s
13	16.5 q	15.9 q	119.9 t
14	24.4 q	22.2 q	22.1 q
15	23.3 q	26.9 q	26.5 q
OCOR	180.6 s	166.8 s	—
	19.6 q	126.9 s	
	19.1 q	140.3 d	
	35.4 d	15.1 q	
		20.6 q	

*Some signals may be interchangeable

†D₃COD

‡Singlets assigned by selective INEPT

ate was most likely as the ^1H NMR data indicated again the presence of a guaianolide with no double bonds. The signals of H-5–H-9, H-11, H-13 and H-14 were again close to those of **2a**. As the couplings of H-8 differed a changed conformation was indicated. All data agreed with the presence of a triol formed by hydrolysis of **2a**. The relative position of the hydroxy groups and the configurations had to be established. Acetylation affo-

rded the monoacetates **7g** and **7h** and the diacetate **7i**. The ^1H NMR data of these derivatives (Table 3) indicated that again the conformations are partly changed. Inspection of models led to the conclusions that all data would agree with the triol **7a** where hydrogen bond between 10-OH and 8-OR was present. Accordingly, in the acetate **7h** the hydrogen bond was missing. This assumption required in part the proposed configurations.

Table 3 ^1H NMR spectral data of **7a-i**, **8a-d**, **9a** and **10** (400 MHz, CDCl_3 , δ -values*)

H	7a	7b†	7e	7f	7g	7h	7i	8a	8b†	9a	10
2	3.68 s	3.68 s	3.68 s	3.71 s	3.65 s	3.78 br s	3.67 s	3.33 d	3.32 d	6.32 d	$\left\{ \begin{array}{l} 2.82 \text{ dd} \\ 2.34 \text{ dd} \end{array} \right.$
3	4.06 s	4.06 s	4.07 s	4.08 s	5.03 s	4.07 s	4.92 s	3.31 d	3.31 d	6.25 d	4.72 br d
5	2.66 d	2.65 d	2.66 d	2.64 d	3.52 d	2.70 d	3.24 d	2.67 d	2.67 d	2.68 d	—
6	4.32 dd	4.30 dd	4.30 dd	4.35 dd	4.38 dd	4.34 dd	4.28 dd	4.49 dd	4.44 dd	3.76 t	4.98 d
7	3.12 q	3.06 q	3.07 q	3.43 m	2.59 q	2.83 q	2.91 q	2.81 q	2.80 q	2.95 q	3.74 q
8	5.26 ddd	5.16 ddd	5.19 ddd	$\left\{ \begin{array}{l} 2.33 \text{ m} \\ 1.61 \text{ m} \end{array} \right.$	5.24 ddd	5.23 br dd	5.22 ddd	5.42 ddd	5.24 ddd	5.14 ddd	6.61 dd
9	2.18 dd	2.15 dd	2.18 dd	2.05 ddd	2.17 dd	2.93 dd	2.98 dd	$\left. \begin{array}{l} 2.00 \text{ m} \\ 2.05 \text{ m} \end{array} \right\}$	2.05 m	2.18 dd	$\left. \begin{array}{l} 6.11 \text{ d} \\ 2.04 \text{ dd} \end{array} \right\}$
9'	2.13 dd	2.05 dd	2.11 dd	1.86 ddd	2.06 dd	2.15 dd	2.16 dd			2.04 dd	
11	2.46 dq	2.44 dq	2.44 dq	$\left\{ \begin{array}{l} 6.21 \text{ d} \\ 5.50 \text{ d} \end{array} \right.$	2.54 dq	2.43 dq	2.44 dq	2.43 dq	2.43 dq	2.34 dq	2.63 dq
13	1.27 d	1.30 d	1.30 d		1.29 d	1.30 d	1.30 d	1.34 d	1.33 d	1.24 d	1.31 d
14	1.25 s	1.25 s	1.26 s	1.28 s	1.41 s	1.57 s	1.50 s	1.51 s	1.51 s	1.36 s	2.27 s
15	1.54 s	1.55 s	1.55 s	1.58 s	1.63 s	1.50 s	1.87 s	1.67 s	1.68 s	1.68 s	2.20 br s

*OCORs Table 1

†Isovalerates and methylbutyrates nearly identical signals

$J[\text{Hz}]$ Compounds **7a-7i** 5,6=6,7=7,8=7,11=11, 8,9=5, 8,9'=3.5, 9,9'=16, 11,13=7 (compounds **7g-7i** 8,9~0.5, 8,9'=6.5, compound **7f** 7,13'=3.5, 7,13'=3, 8,9=8,9'=6.5, 8,9=3.5, 8,9'=9.5, 9,9'=14), compounds **8a-d** 2,3=3.5, 5,6=6,7=7,8=7,11~11, 8,9=8,9'=11,13=7, compound **9a** 2,3=5.5, 5,6=6,7=7,8=7,11~11, 8,9=7, 8,9'=2.5, 9,9'=16.5, 11,13=7, compound **10** 2,2'=18, 2,3=6, 2',3=2.5, 6,7=9.5, 7,8=8.5, 7,11=11.5, 8,9=16.5, 11,13=7

The whole stereochemistry was established by NOE difference spectroscopy in a chloroform-benzene mixture where also the hydroxy signals were visible. Clear effects were obtained between H-3, 2-OH (10%), H-5 (4%) and H-15 (4%), between H-2, H-3 (15%) and H-14 (8%), between 4-OH, H-3 (10%) and H-5 (15%), between H-5 and H-7 (10%), between H-13 and H-7 (8%), between H-15 and H-6 (8%), between H-14, H-6 (4%), H-2 (10%) and H-9 β (8%) as well as between H-8, H-6 (8%) and H-11 (12%). Also the ^{13}C NMR spectrum (Table 2) supported the structure. Selective irradiation of H-14 and H-15 respectively allowed the assignment of the signals for C-10 and C-4. The chemical shift of the latter (δ 83.1) excluded the presence of a 3,4-epoxide. The ^1H NMR spectra of **7b-e** (Table 3) clearly indicated the presence of the corresponding isobutyrate, isovalerate, methylbutyrate and propionate. The configuration of **7a** agrees with the expected one if **2a** was the precursor.

A closely related guaianolide (**7f**) was isolated from *Pentzia albidia* (DC) Huzel var *annua* (DC) Merx et Eberle together with artabin, tanaparthin-1 α ,4 α -peroxide [7], nerolidol and the derivatives **13a-c** [12]. The ^1H NMR spectrum of **7f** (Table 3) differed from that of **7a** by the absence of ester signals and the presence of exomethylene proton signals. The low field signal for H-8 was replaced by a pair of multiplets at 2.33 and 1.61 and the H-9 signals were threefold doublets. Thus **7f** was the 11,13-dehydro-8-deacyloxy derivative of **7a**. Identical stereochemistry was established by the observed NOE's which were analogous with those of **7a**. Also the ^{13}C NMR spectrum (Table 2) agreed with the structure.

The ^1H NMR spectrum of **8b** (Table 3) was in part close to that of **4b**. However, the signals of the olefinic protons were replaced by a AB quartet at 3.31, obviously due to epoxide protons. Accordingly, the molecular formula was $\text{C}_{19}\text{H}_{28}\text{O}_8$ differing from that of **4b** by one oxygen. As the chemical shift of H-5 was nearly the same as in the case of **4b** the presence of a 2,3 β -epoxide was

very likely. The ^1H NMR spectra of **8a** and **8c/d** (Table 3) indicated that again also the corresponding angelate, isovalerate and methylbutyrate were present.

The structure of **9a** could be easily deduced from its ^1H NMR spectrum which was very close to those of **9b-d** [10]. The presence of the corresponding angelate followed from the typical signals. The endoperoxides **9a-9d** surely are the precursors of the bis-epoxides **3a-3d**. These lactones were not isolated from the material from the Botanical Garden Berlin. However, only a small sample was investigated.

The molecular formula of **10** ($\text{C}_{15}\text{H}_{18}\text{O}_5$) and the ^1H NMR spectrum (Table 3) indicated the presence of a seco-guaianolide. The relative position of the oxygen functions followed from the results of spin decoupling and the chemical shifts. Most likely a guaianolide with oxygen functions at C-3, C-4 and C-8 and a 1(10)double bond is a precursor of **10** which may be formed by oxidative degradation followed by elimination.

The structures of **11a** and **11b** followed from the ^1H NMR spectra (Table 4) which were close to that of artemisia alcohol. The relative position of the second hydroxy group followed from the shift differences of H-5, H-7 and H-8. The ^1H NMR spectrum of **12** (Table 4) was close to that of the epimeric diol [13]. The changed configuration at C-5 followed from the absence of vicinal couplings.

The isolation of a large variety of highly oxygenated guaianolides from *Otanithus maritimus* is of chemotaxonomic interest and indicates a relationship to *Artemisia* where in part very similar lactones are common. The isolation of **7f** from a *Pentzia* species agrees with previous results which also may indicate a relation of this genus to *Artemisia*.

EXPERIMENTAL

The air-dried aerial parts (8 kg, collected in May 1986 in sandy coasted strips, West of Alexandria, identified by Prof. Dr

Table 4 ^1H NMR spectral data of **11a**, **11b** and **12** (400 MHz, CDCl_3 , δ -values)

	11a	11b	12
1	{ 5.15 dd 5.10 dd (t)	{ 5.13 dd 5.09 dd	—
2	5.86 dd	5.89 dd	4.38 br d
3	—	—	{ 2.33 ddd 1.91 br dd
4	4.14 d	4.11 d	2.05 ddd
5	5.39 br d	5.47 br d	4.45 s
6	—	—	2.58 d
7	1.85 d	4.05 br s	{ 5.05 br s 4.87 br s
8	{ 4.27 d 4.05 d	1.73 br s	—
9	1.02 s	1.02 s	1.18 s
10	1.01 s	1.01 s	1.33 s

$J[\text{Hz}]$ Compounds **11a** and **b** 1t,2=17, 1c,2=10, 1r,1c=1, 4,5=9, 5,7 or 5,8=1.5, compound **12** 2,3=8, 3,4=3, 3',4=3.5, 3,3'=14, 4,5=6

V. Tackholm, Cairo University) was extracted with Et_2O -petrol, 1:1. The extract obtained (375 g) was filtered over a silica gel column and the polar CC fractions (SiO_2 , Et_2O and Et_2O -MeOH) were separated again by CC into four crude fractions (1 Et_2O -Petrol, 1:1, 2 Et_2O -Petrol, 3:1; 3 Et_2O and 4 Et_2O -MeOH, 9:1). One third of fraction 1 (0.3 g) was further separated by TLC (Et_2O -petrol, 3:1) into three bands (1/1-1/3) HPLC of 1/1 (MeOH- H_2O , 7:3, always RP 8, ca 100 bar) afforded 5 mg undeca-2,4-diene-8,10-dynoic acid isobutyl amide (R_f 9.0 min), 10 mg **8a** (R_f 10.8 min), 3 mg **8b** (R_f 10.3 min) and 8 mg **8c/d** (ca 2:3, R_f 11.5 min) HPLC of 1/2 (MeOH- H_2O , 3:1) afforded 15 mg **7a**, mp 232° (R_f 9.9 min), 5 mg **7b** (R_f 9.5 min), 10 mg **7c/d** (R_f 10.5 min) HPLC of 1/3 (MeOH- H_2O , 1:1) gave 4 mg **11a** (R_f 8.8 min), 8 mg **11b** (R_f 7.5 min), and 10 mg **12** (R_f 6.5 min) TLC (Et_2O -petrol, 3:1) and HPLC (MeOH- H_2O , 3:2) of 5% (100 mg) of SC 2 gave 15 mg artabsin, 8 mg lololide, 10 mg sesamin 5 mg **1a** (R_f 10.4 min), 5 mg **16** (R_f 10.0 min), 10 mg **1c/d** (R_f 11.0 min) and 5 mg **7e** (R_f 10.6 min) Repeated HPLC of 20% of SC 3 (300 mg) (MeOH- H_2O , 1:1) afforded 12 mg **3a** (R_f 4.8 min), 12 mg **3b** (R_f 5.2 min), 10 mg **3c/d** (R_f 5.6 min), 15 mg romalea allene (R_f 6.0 min), 12 mg vomifoliol (R_f 6.5 min), 7 mg **4a** (R_f 12.3 min), 9 mg **4c/d** (R_f 13.0 min), 5 mg **5b** (R_f 13.5 min), 3 mg **5a** (R_f 14.0 min), 5 mg **5c/d** (R_f 14.8 min),

Table 5 Mass spectral data of the guaianolides, **11a/b** and **12**

1a	360 157 $[\text{M}]^+$ (6) ($\text{C}_{20}\text{H}_{24}\text{O}_6$), 260 $[\text{M}-\text{RCO}_2\text{H}]^+$ (9), 83 $[\text{RCO}]^+$ (100)
1b	348 $[\text{M}]^+$ (2.5), 260 $[\text{M}-\text{RCO}_2\text{H}]^+$ (8), 71 $[\text{RCO}]^+$ (100)
1c/d	362 $[\text{M}]^+$ (6), 260 $[\text{M}-\text{RCO}_2\text{H}]^+$ (10), 85 $[\text{RCO}]^+$ (44), 57 $[\text{85}-\text{CO}]^+$ (100)
2a	378 $[\text{M}]^+$ (0.2), 360 $[\text{M}-\text{H}_2\text{O}]^+$ (0.3), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (0.7), 83 $[\text{RCO}]^+$ (100)
2b	366 $[\text{M}]^+$ (0.1), 358 $[\text{M}-\text{H}_2\text{O}]^+$ (0.2), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (1), 71 $[\text{RCO}]^+$ (100)
2c/d	380 $[\text{M}]^+$ (0.2), 362 $[\text{M}-\text{H}_2\text{O}]^+$ (0.2), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (1), 85 $[\text{RCO}]^+$ (45), 57 $[\text{85}-\text{CO}]^+$ (100)
3a	378 $[\text{M}]^+$ (0.1), 360 $[\text{M}-\text{H}_2\text{O}]^+$ (0.2), 318 $[\text{360}-\text{C}_2\text{H}_2\text{O}]^+$ (1.5), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (1), 218 $[\text{318}-\text{RCO}_2\text{H}]^+$ (62), 83 $[\text{RCO}]^+$ (100)
3b	366 167 $[\text{M}]^+$ (0.1) ($\text{C}_{19}\text{H}_{26}\text{O}_7$), 348 $[\text{M}-\text{H}_2\text{O}]^+$ (0.2), 306 $[\text{348}-\text{C}_2\text{H}_2\text{O}]^+$ (4.5), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (1), 218 $[\text{306}-\text{RCO}_2\text{H}]^+$ (48), 71 $[\text{C}_3\text{H}_7\text{CO}]^+$ (100)
3c/d	380 $[\text{M}]^+$ (0.1), 362 $[\text{M}-\text{H}_2\text{O}]^+$ (0.2), 320 $[\text{362}-\text{C}_2\text{H}_2\text{O}]^+$ (2.5), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (1), 218 $[\text{320}-\text{RCO}_2\text{H}]^+$ (56), 85 $[\text{RCO}]^+$ (52), 57 $[\text{85}-\text{CO}]^+$ (100)
4a	362 $[\text{M}-\text{H}_2\text{O}]^+$ (0.3), 262 $[\text{362}-\text{RCO}_2\text{H}]^+$ (8), 83 $[\text{RCO}]^+$ (100)
4b	353 160 $[\text{M}-\text{Me}]^+$ (7) ($\text{C}_{18}\text{H}_{25}\text{O}_7$), 350 $[\text{M}-\text{H}_2\text{O}]^+$ (0.6), 332 $[\text{350}-\text{H}_2\text{O}]^+$ (1.5), 262 $[\text{350}-\text{RCO}_2\text{H}]^+$ (10), 167 $[\text{C}_9\text{H}_{11}\text{O}_3]^+$ (100), 71 $[\text{C}_3\text{H}_7\text{CO}]^+$ (96)
4c/d	364 189 $[\text{M}-\text{H}_2\text{O}]^+$ (0.5) ($\text{C}_{20}\text{H}_{28}\text{O}_6$), 262 $[\text{364}-\text{RCO}_2\text{H}]^+$ (7), 244 $[\text{262}-\text{H}_2\text{O}]^+$ (4), 85 $[\text{RCO}]^+$ (45), 57 $[\text{85}-\text{CO}]^+$ (100)
5a	378 $[\text{M}-\text{HCl}]^+$ (0.1) ($\text{C}_{20}\text{H}_{27}\text{O}_7$), 278 $[\text{378}-\text{RCO}_2\text{H}]^+$ (1.3), 260 $[\text{278}-\text{H}_2\text{O}]^+$ (5), 218 $[\text{260}-\text{C}_2\text{H}_2\text{O}]^+$ (56), 83 $[\text{C}_4\text{H}_7\text{CO}]^+$ (100)
5b	402 $[\text{M}]^+$ (0.3), 366 167 $[\text{M}-\text{HCl}]^+$ (0.8) ($\text{C}_{19}\text{H}_{26}\text{O}_7$), 348 $[\text{366}-\text{H}_2\text{O}]^+$ (1.3), 333 $[\text{348}-\text{Me}]^+$ (3), 306 $[\text{348}-\text{C}_2\text{H}_2\text{O}]^+$ (65), 218 $[\text{306}-\text{RCO}_2\text{H}]^+$ (74), 71 $[\text{C}_3\text{H}_7\text{CO}]^+$ (100)
5c/d	380 $[\text{M}-\text{HCl}]^+$ (0.1), 278 $[\text{380}-\text{RCO}_2\text{H}]^+$ (1), 85 $[\text{RCO}]^+$ (46), 57 $[\text{85}-\text{CO}]^+$ (100)
6a	378 $[\text{M}-\text{H}_2\text{O}]^+$ (0.1), 318 (2), 278 (1), 218 (42), 83 (100)
6b	366 167 $[\text{M}-\text{H}_2\text{O}]^+$ (0.5) ($\text{C}_{19}\text{H}_{26}\text{O}_7$), 348 $[\text{366}-\text{H}_2\text{O}]^+$ (1), 306 $[\text{348}-\text{C}_2\text{H}_2\text{O}]^+$ (7), 278 $[\text{366}-\text{RCO}_2\text{H}]^+$ (2), 218 $[\text{306}-\text{RCO}_2\text{H}]^+$ (66), 71 $[\text{RCO}]^+$ (100)
7a	396 $[\text{M}]^+$ (1) ($\text{C}_{20}\text{H}_{28}\text{O}_8$), 381 $[\text{M}-\text{Me}]^+$ (0.4), 363 $[\text{381}-\text{H}_2\text{O}]^+$ (2.5), 83 $[\text{C}_4\text{H}_7\text{CO}]^+$ (100)
7b	384 178 $[\text{M}]^+$ (1) ($\text{C}_{19}\text{H}_{28}\text{O}_8$), 71 $[\text{C}_3\text{H}_7\text{CO}]^+$ (100)
7c/d	398 $[\text{M}]^+$ (0.6) ($\text{C}_{20}\text{H}_{30}\text{O}_8$), 383 $[\text{M}-\text{Me}]^+$ (0.6), 85 $[\text{C}_4\text{H}_6\text{CO}]^+$ (68), 57 $[\text{85}-\text{CO}]^+$ (100)
7e	370.163 $[\text{M}]^+$ (0.6) ($\text{C}_{18}\text{H}_{26}\text{O}_8$), 296 $[\text{M}-\text{RCO}_2\text{H}]^+$ (1.3), 279 $[\text{296}-\text{OH}]^+$ (14), 261 $[\text{279}-\text{H}_2\text{O}]^+$ (7), 57 $[\text{RCO}]^+$ (100)
7f	(CIMS) 297 $[\text{M}+1]^+$ (20), 269 $[\text{297}-\text{H}_2\text{O}]^+$ (19), 57 (100)
8a	394 162 $[\text{M}]^+$ (1.1) ($\text{C}_{20}\text{H}_{26}\text{O}_8$), 294 $[\text{M}-\text{RCO}_2\text{H}]^+$ (9), 167 (80), 83 $[\text{RCO}]^+$ (100), 55 $[\text{83}-\text{CO}]^+$ (60)
8b	382 178 $[\text{M}]^+$ (4) ($\text{C}_{19}\text{H}_{28}\text{O}_8$), 294 $[\text{M}-\text{RCO}_2\text{H}]^+$ (9), 167 (100), 71 $[\text{RCO}]^+$ (70)
8c/d	396 178 $[\text{M}]^+$ (1.7) ($\text{C}_{20}\text{H}_{28}\text{O}_8$), 294 $[\text{M}-\text{RCO}_2\text{H}]^+$ (10), 167 (82), 85 $[\text{RCO}]^+$ (60), 57 $[\text{85}-\text{CO}]^+$ (100)
9a	378 168 $[\text{M}]^+$ (0.5) ($\text{C}_{20}\text{H}_{26}\text{O}_7$), 346 $[\text{M}-\text{O}_2]^+$ (0.5), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (6), 260 $[\text{278}-\text{H}_2\text{O}]^+$ (7), 111 (58), 83 $[\text{RCO}]^+$ (80), 55 $[\text{83}-\text{CO}]^+$ (100)
9b	366 168 $[\text{M}]^+$ (2) ($\text{C}_{19}\text{H}_{26}\text{O}_7$), 334 $[\text{M}-\text{O}_2]^+$ (3), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (8), 111 (100), 71 $[\text{RCO}]^+$ (88)
9c/d	380 $[\text{M}]^+$ (0.6) ($\text{C}_{20}\text{H}_{28}\text{O}_7$), 348 $[\text{M}-\text{O}_2]^+$ (0.3), 278 $[\text{M}-\text{RCO}_2\text{H}]^+$ (6.5), 85 $[\text{RCO}]^+$ (64), 57 $[\text{85}-\text{CO}]^+$ (100)
10	278 115 $[\text{M}]^+$ (10) ($\text{C}_{15}\text{H}_{18}\text{O}_5$), 260 $[\text{M}-\text{H}_2\text{O}]^+$ (10), 95 (100)
11a	(CIMS) 171 $[\text{M}+1]^+$ (2), 153 $[\text{171}-\text{H}_2\text{O}]^+$ (86), 135 $[\text{153}-\text{H}_2\text{O}]^+$ (100)
11b	101 $[\text{M}-\text{C}_3\text{H}_9]^+$ (56), 83 $[\text{101}-\text{H}_2\text{O}]^+$ (80), 55 $[\text{83}-\text{CO}]^+$ (100)

15 mg **2b** (R_f 20.5 min), 15 mg **2a** (R_f 21.0 min), 8 mg **2c/d** (R_f 22.0 min), 3 mg **6b** (R_f 23.0 min), 3 mg **6a** (R_f 24.5 min) and 2 mg **6c/d** (R_f 26.0 min) HPLC of 5% of SC 4 (100 mg) (MeOH–H₂O, 1:1) gave 8 mg **4b** (R_f 8.0 min) and 4 mg **10** (R_f 4.5 min). 80 g aerial parts grown in the Botanical Garden Berlin gave a polar SC fraction which afforded by HPLC (MeOH–H₂O, 3:2) 6 mg **9b** (R_f 6.1 min), 4 mg **9a** (R_f 8.2 min), 6 mg **9c/d** (R_f 8.8 min) and 7 mg tanaparthin-1 α ,4 α -peroxide.

The extract of the aerial parts of *Pentzia albidia* (110 g, voucher M. Müller 3704, collected at the Luderitz Bay) gave two crude SC fractions (SC 1 and 2). TLC of SC 1 (Et₂O–petrol, 1:1) gave 30 mg nerolidol and of SC 2 (Et₂O–MeOH, 9:1) 5 mg **13c**, 5 mg **13a**, 10 mg **13b**, 8 mg tanaparthin-1 α ,4 α -peroxide, 45 mg artabin and a mixture which gave by HPLC (MeOH–H₂O, 7:3) 5 mg artabin and 5 mg **7f** (R_f 9.5 min). Known compounds were identified by comparing the 400 MHz ¹H NMR spectra with those of authentic material. The mass spectral data of the new compounds are summarized in Table 5.

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